Signature of Invigilators

1. .....

2. .....

Roll No. | | | | | |

(In figures as in Admit Card)

CHEMICAL SCIENCES

Paper III

Roll No. .....

.....

(In words)

J—0302

Name of Areas/Section (if any) .....

Time Allowed : $2\frac{1}{2}$ Hours] [Maximum Marks : 200

Instructions for the Candidates

1. Write your Roll number in the space provided on the top of this page.
2. Write name of your Elective/Section if any.
3. Answer to short answer/essay type questions are to be written in the space provided below each question or after the questions in test booklet itself. No additional sheets are to be used.
4. Read instructions given inside carefully.
5. Last page is attached at the end of the test booklet for rough work.
6. If you write your name or put any special mark on any part of the test booklet which may disclose in any way your identity, you will render yourself liable to disqualification.
7. Use of any calculator is prohibited.
8. There is no negative marking.
9. You should return the test booklet to the invigilator at the end of the examination and should not carry any paper outside the examination hall.

પરીક્ષાર્થીઓ માટેની સૂચનાઓ :

૧. આ પાનાની ટોચમાં દર્શાવેલી જગ્યામાં તમારો રોલ નંબર લખો.

૨. જો કોઈ વિકલ્પ/વિભાગ પસંદ કર્યા હોય તો તે યોગ્ય જગ્યાએ દર્શાવો.

૩. ટુંકા પ્રશ્નો/નિબંધ વિષેના જવાબો એ પ્રશ્નની નીચે અથવા બાજુમાં આપેલી જગ્યામાં લખો. વધારાના કોઈ પાનાનો ઉપયોગ કરશો નહીં.

૪. અંદર આપેલી સૂચનાઓ કાળજીપૂર્વક વાંચો.

૫. બુકલેટની પાછળ આપેલું છેલ્લું પાનું રફ કામ માટે છે.

૬. બુકલેટ કોઈપણ ઠેકાણે તમારું નામ કે કોઈ ચોકકસ સંજ્ઞા કરવી નહીં કે જે તમારી ઓળખ પૂરી પાડે. આ તમને પરીક્ષા માટે ગેરલાયક ઠેરવશે.

૭. કેલ્કયુલેટર નો ઉપયોગ કરાશે નહી.

૮. નકારાત્મક માર્કીંગ નથી.

૯. પરીક્ષા સમય પૂરો થઈ ગયા પછી આ બુકલેટ જે તે નીરીક્ષકને સોંપી દેવી. કોઈપણ પેપર પરીક્ષા ખંડની બહાર લઈ જવું નહી.

FOR OFFICE USE ONLY
Marks Obtained

Question Number	Marks Obtained	Question Number	Marks Obtained	Question Number	Marks Obtained
1		26			
2		27			
3		28			
4		29			
5		30			
6		31			
7		32			
8		33			
9		34			
10		35			
11		36			
12					
13					
14					
15					
16					
17					
18					
19					
20					
21					
22					
23					
24					
5					

Total Marks Obtained.....

Signature of the co-ordinator.....
(Evaluation)

	0	1	2	3	4	5	6	7	8	9	1 2 3	4 5 6	7 8 9
·00	1000	1002	1005	1007	1009	1012	1014	1016	1019	1021	0 0 1	1 1 1	2 2 2
·01	1023	1026	1028	1030	1033	1035	1038	1040	1042	1045	0 0 1	1 1 1	2 2 2
·02	1047	1050	1052	1054	1057	1059	1062	1064	1067	1069	0 0 1	1 1 1	2 2 2
·03	1072	1074	1076	1079	1081	1084	1086	1089	1091	1094	0 0 1	1 1 1	2 2 2
·04	1096	1099	1102	1104	1107	1109	1112	1114	1117	1119	0 1 1	1 1 2	2 2 2
·05	1122	1125	1127	1130	1132	1135	1138	1140	1143	1146	0 1 1	1 1 2	2 2 2
·06	1148	1151	1153	1156	1159	1161	1164	1167	1169	1172	0 1 1	1 1 2	2 2 2
·07	1175	1178	1180	1183	1186	1189	1191	1194	1197	1199	0 1 1	1 1 2	2 2 2
·08	1202	1205	1208	1211	1213	1216	1219	1222	1225	1227	0 1 1	1 1 2	2 2 3
·09	1230	1233	1236	1239	1242	1245	1247	1250	1253	1256	0 1 1	1 1 2	2 2 3
·10	1259	1262	1265	1268	1271	1274	1276	1279	1282	1285	0 1 1	1 1 2	2 2 3
·11	1288	1291	1294	1297	1300	1303	1306	1309	1312	1315	0 1 1	1 2 2	2 2 3
·12	1318	1321	1324	1327	1330	1334	1337	1340	1343	1346	0 1 1	1 2 2	2 2 3
·13	1349	1352	1355	1358	1361	1365	1368	1371	1374	1377	0 1 1	1 2 2	2 3 3
·14	1380	1384	1387	1390	1393	1396	1400	1403	1406	1409	0 1 1	1 2 2	2 3 3
·15	1413	1416	1419	1422	1426	1429	1432	1435	1439	1442	0 1 1	1 2 2	2 3 3
·16	1445	1449	1452	1455	1459	1462	1466	1469	1472	1476	0 1 1	1 2 2	2 3 3
·17	1479	1483	1486	1489	1493	1496	1500	1503	1507	1510	0 1 1	1 2 2	2 3 3
18	1514	1517	1521	1524	1528	1531	1535	1538	1542	1545	0 1 1	1 2 2	2 3 3
19	1549	1552	1556	1560	1563	1567	1570	1574	1578	1581	0 1 1	1 2 2	3 3 3
·20	1585	1589	1592	1596	1600	1603	1607	1611	1614	1618	0 1 1	1 2 2	3 3 3
21	1622	1626	1629	1633	1637	1641	1644	1648	1652	1656	0 1 1	2 2 2	3 3 3
22	1660	1663	1667	1671	1675	1679	1683	1687	1690	1694	0 1 1	2 2 2	3 3 3
·23	1698	1702	1706	1710	1714	1718	1722	1726	1730	1734	0 1 1	2 2 2	3 3 4
24	1738	1742	1746	1750	1754	1758	1762	1766	1770	1774	0 1 1	2 2 2	3 3 4
·25	1778	1782	1786	1791	1795	1799	1803	1807	1811	1816	0 1 1	2 2 2	3 3 4
26	1820	1824	1828	1832	1837	1841	1845	1849	1854	1858	0 1 1	2 2 3	3 3 4
27	1862	1866	1871	1875	1879	1884	1888	1892	1897	1901	0 1 1	2 2 3	3 3 4
28	1905	1910	1914	1919	1923	1928	1932	1936	1941	1945	0 1 1	2 2 3	3 4 4
29	1950	1954	1959	1963	1968	1972	1977	1982	1986	1991	0 1 1	2 2 3	3 4 4
·30	1995	2000	2004	2009	2014	2018	2023	2028	2032	2037	0 1 1	2 2 3	3 4 4
31	2042	2046	2051	2056	2061	2065	2070	2075	2080	2084	0 1 1	2 2 3	3 4 4
32	2089	2094	2099	2104	2109	2113	2118	2123	2128	2133	0 1 1	2 2 3	3 4 4
33	2138	2143	2148	2153	2158	2163	2168	2173	2178	2183	0 1 1	2 2 3	3 4 4
34	2188	2193	2198	2203	2208	2213	2218	2223	2228	2234	1 1 2	2 3 3	4 4 5
·35	2239	2244	2249	2254	2259	2265	2270	2275	2280	2286	1 1 2	2 3 3	4 4 5
36	2291	2296	2301	2307	2312	2317	2323	2328	2333	2339	1 1 2	2 3 3	4 4 5
37	2344	2350	2355	2360	2366	2371	2377	2382	2388	2393	1 1 2	2 3 3	4 4 5
38	2399	2404	2410	2415	2421	2427	2432	2438	2443	2449	1 1 2	2 3 3	4 4 5
39	2455	2460	2466	2472	2477	2483	2489	2495	2500	2506	1 1 2	2 3 3	4 5 5
·40	2512	2518	2523	2529	2535	2541	2547	2553	2559	2564	1 1 2	2 3 4	4 5 5
41	2570	2576	2582	2588	2594	2600	2606	2612	2618	2624	1 1 2	2 3 4	4 5 5
42	2630	2636	2642	2649	2655	2661	2667	2673	2679	2685	1 1 2	2 3 4	4 5 6
43	2692	2698	2704	2710	2716	2723	2729	2735	2742	2748	1 1 2	3 3 4	4 5 6
44	2754	2761	2767	2773	2780	2786	2793	2799	2805	2812	1 1 2	3 3 4	4 5 6
·45	2818	2825	2831	2838	2844	2851	2858	2864	2871	2877	1 1 2	3 3 4	5 5 6
46	2884	2891	2897	2904	2911	2917	2924	2931	2938	2944	1 1 2	3 3 4	5 5 6
47	2951	2958	2965	2972	2979	2985	2992	2999	3006	3013	1 1 2	3 3 4	5 5 6
48	3020	3027	3034	3041	3048	3055	3062	3069	3076	3083	1 1 2	3 4 4	5 6 6
49	3090	3097	3105	3112	3119	3126	3133	314	3148	3155	1 1 2	3 4 4	5 6 6

	0	1	2	3	4	5	6	7	8	9	1 2 3	4 5 6	7 8 9
·50	3162	3170	3177	3184	3192	3199	3206	3214	3221	3228	1 1 2	3 4 4	5 6 7
51	3236	3243	3251	3258	3266	3273	3281	3289	3296	3304	1 2 2	3 4 5	5 6 7
52	3311	3319	3327	3334	3342	3350	3357	3365	3373	3381	1 2 2	3 4 5	5 6 7
53	3388	3396	3404	3412	3420	3428	3436	3443	3451	3459	1 2 2	3 4 5	6 6 7
54	3467	3475	3483	3491	3499	3508	3516	3524	3532	3540	1 2 2	3 4 5	6 6 7
·55	3548	3556	3565	3573	3581	3589	3597	3606	3614	3622	1 2 2	3 4 5	6 7 7
56	3631	3639	3648	3656	3664	3673	3681	3690	3698	3707	1 2 3	3 4 5	6 7 8
57	3715	3724	3733	3741	3750	3758	3767	3776	3784	3793	1 2 3	3 4 5	6 7 8
58	3802	3811	3819	3828	3837	3846	3855	3864	3873	3882	1 2 3	4 4 5	6 7 8
59	3890	3899	3908	3917	3926	3936	3945	3954	3963	3972	1 2 3	4 5 5	6 7 8
·60	3981	3990	3999	4009	4018	4027	4036	4046	4055	4064	1 2 3	4 5 6	6 7 8
61	4074	4083	4093	4102	4111	4121	4130	4140	4150	4159	1 2 3	4 5 6	7 8 9
62	4169	4178	4188	4198	4207	4217	4227	4236	4246	4256	1 2 3	4 5 6	7 8 9
63	4266	4276	4285	4295	4305	4315	4325	4335	4345	4355	1 2 3	4 5 6	7 8 9
64	4365	4375	4385	4395	4406	4416	4426	4436	4446	4457	1 2 3	4 5 6	7 8 9
·65	4467	4477	4487	4498	4508	4519	4529	4539	4550	4560	1 2 3	4 5 6	7 8 9
66	4571	4581	4592	4603	4613	4624	4634	4645	4656	4667	1 2 3	4 5 6	7 9 10
67	4677	4688	4699	4710	4721	4732	4742	4753	4764	4775	1 2 3	4 5 7	8 9 10
68	4786	4797	4808	4819	4831	4842	4853	4864	4875	4887	1 2 3	4 6 7	8 9 10
69	4898	4909	4920	4932	4943	4955	4966	4977	4989	5000	1 2 3	5 6 7	8 9 10
·70	5012	5023	5035	5047	5058	5070	5082	5093	5105	5117	1 2 4	5 6 7	8 9 11
71	5129	5140	5152	5164	5176	5188	5200	5212	5224	5236	1 2 4	5 6 7	8 10 11
72	5248	5260	5272	5284	5297	5309	5321	5333	5346	5358	1 2 4	5 6 7	9 10 11
73	5370	5383	5395	5408	5420	5433	5445	5458	5470	5483	1 3 4	5 6 8	9 10 11
74	5495	5508	5521	5534	5546	5559	5572	5585	5598	5610	1 3 4	5 6 8	9 10 12
·75	5623	5636	5649	5662	5675	5689	5702	5715	5728	5741	1 3 4	5 7 8	9 10 12
76	5754	5768	5781	5794	5808	5821	5834	5848	5861	5875	1 3 4	5 7 8	9 11 12
77	5888	5902	5916	5929	5943	5957	5970	5984	5998	6012	1 3 4	5 7 8	10 11 12
78	6026	6039	6053	6067	6081	6095	6109	6124	6138	6152	1 3 4	6 7 8	10 11 13
79	6166	6180	6194	6209	6223	6237	6252	6266	6281	6295	1 3 4	6 7 9	10 11 13
·80	6310	6324	6339	6353	6368	6383	6397	6412	6427	6442	1 3 4	6 7 9	10 12 13
81	6457	6471	6486	6501	6516	6531	6546	6561	6577	6592	2 3 5	6 8 9	11 12 14
82	6607	6622	6637	6653	6668	6683	6699	6714	6730	6745	2 3 5	6 8 9	11 12 14
83	6761	6776	6792	6808	6823	6839	6855	6871	6887	6902	2 3 5	6 8 9	11 13 14
84	6918	6934	6950	6966	6982	6998	7015	7031	7047	7063	2 3 5	6 8 10	11 13 15
·85	7079	7096	7112	7129	7145	7161	7178	7194	7211	7228	2 3 5	7 8 10	12 13 15
86	7244	7261	7278	7295	7311	7328	7345	7362	7379	7396	2 3 5	7 8 10	12 13 15
87	7413	7430	7447	7464	7482	7499	7516	7534	7551	7568	2 3 5	7 9 10	12 14 16
88	7586	7603	7621	7638	7656	7674	7691	7709	7727	7745	2 4 5	7 9 11	12 14 16
89	7762	7780	7798	7816	7834	7852	7870	7889	7907	7925	2 4 5	7 9 11	13 14 16
·90	7943	7962	7980	7998	8017	8035	8054	8072	8091	8110	2 4 6	7 9 11	13 15 17
91	8128	8147	8166	8185	8204	8222	8241	8260	8279	8299	2 4 6	8 9 11	13 15 17
92	8318	8337	8356	8375	8395	8414	8433	8453	8472	8492	2 4 6	8 10 12	14 15 17
93	8511	8531	8551	8570	8590	8610	8630	8650	8670	8690	2 4 6	8 10 12	14 16 18
94	8710	8730	8750	8770	8790	8810	8831	8851	8872	8892	2 4 6	8 10 12	14 16 18
·95	8913	8933	8954	8974	8995	9016	9036	9057	9078	9099	2 4 6	8 10 12	15 17 19
96	9120	9141	9162	9183	9204	9226	9247	9268	9290	9311	2 4 6	8 11 13	15 17 19
97	9333	9354	9376	9397	9419	9441	9462	9484	9506	9528	2 4 7	9 11 13	15 17 20
98	9550	9572	9594	9616	9638	9661	9683	9705	9727	9750	2 4 7	9 11 13	16 18 20
99	9772	9795	9817	9840	9863	9886	9908	9931	9954	9977	2 5 7	9 11 14	16 18 20

	0	1	2	3	4	5	6	7	8	9	1 2 3	4 5 6	7 8 9
10	0000	0043	0086	0128	0170	0212	0253	0294	0334	0374	5 9 13 4 8 12	17 21 26 16 20 24	30 34 38 28 32 36
11	0414	0453	0492	0531	0569	0607	0645	0682	0719	0755	4 8 12 4 7 11	16 20 23 15 18 22	27 31 35 26 29 33
12	0792	0828	0864	0899	0934	0969	1004	1038	1072	1106	3 7 11 3 7 10	14 18 21 14 17 20	25 28 32 24 27 31
13	1139	1173	1206	1239	1271	1303	1335	1367	1399	1430	3 6 10 3 7 10	13 16 19 13 16 19	23 26 29 22 25 29
14	1461	1492	1523	1553	1584	1614	1644	1673	1703	1732	3 6 9 3 6 9	12 15 19 12 14 17	22 25 28 20 23 26
15	1761	1790	1818	1847	1875	1903	1931	1959	1987	2014	3 6 9 3 6 8	11 14 17 11 14 17	20 23 26 19 22 25
16	2041	2068	2095	2122	2148	2175	2201	2227	2253	2279	3 6 8 3 5 8	11 14 16 10 13 16	19 22 24 18 21 23
17	2304	2330	2355	2380	2405	2430	2455	2480	2504	2529	3 5 8 3 5 8	10 13 15 10 12 15	18 20 23 17 20 22
18	2553	2577	2601	2625	2648	2672	2695	2718	2742	2765	2 5 7 2 4 7	9 12 14 9 11 14	17 19 21 16 18 21
19	2788	2810	2833	2856	2878	2900	2923	2945	2967	2989	2 4 7 2 4 6	9 11 13 8 11 13	16 18 20 15 17 19
20	3010	3032	3054	3075	3096	3118	3139	3160	3181	3201	2 4 6	8 11 13	15 17 19
21	3222	3243	3263	3284	3304	3324	3345	3365	3385	3404	2 4 6	8 10 12	14 16 18
22	3424	3444	3464	3483	3502	3522	3541	3560	3579	3598	2 4 6	8 10 12	14 15 17
23	3617	3636	3655	3674	3692	3711	3729	3747	3766	3784	2 4 6	7 9 11	13 15 17
24	3802	3820	3838	3856	3874	3892	3909	3927	3945	3962	2 4 5	7 9 11	12 14 16
25	3979	3997	4014	4031	4048	4065	4082	4099	4116	4133	2 3 5	7 9 10	12 14 15
26	4150	4166	4183	4200	4216	4232	4249	4265	4281	4298	2 3 5	7 8 10	11 13 15
27	4314	4330	4346	4362	4378	4393	4409	4425	4440	4456	2 3 5	6 8 9	11 13 14
28	4472	4487	4502	4518	4533	4548	4564	4579	4594	4609	2 3 5	6 8 9	11 12 14
29	4624	4639	4654	4669	4683	4698	4713	4728	4742	4757	1 3 4	6 7 9	10 12 13
30	4774	4786	4800	4814	4829	4843	4857	4871	4886	4900	1 3 4	6 7 9	10 11 13
31	4914	4928	4942	4955	4969	4983	4997	5011	5024	5038	1 3 4	5 7 8	10 11 12
32	5051	5065	5079	5092	5105	5119	5132	5145	5159	5172	1 3 4	5 7 8	9 11 12
33	5185	5198	5211	5224	5237	5250	5263	5276	5289	5302	1 3 4	5 6 8	9 10 12
34	5315	5328	5340	5353	5366	5378	5391	5403	5416	5428	1 3 4	5 6 8	9 10 11
35	5441	5453	5465	5478	5490	5502	5514	5527	5539	5551	1 2 4	5 6 7	9 10 11
36	5563	5575	5587	5599	5611	5623	5635	5647	5658	5670	1 2 4	5 6 7	8 10 11
37	5682	5694	5705	5717	5729	5740	5752	5763	5775	5786	1 2 3	5 6 7	8 9 10
38	5798	5809	5821	5832	5843	5855	5866	5877	5888	5899	1 2 3	5 6 7	8 9 10
39	5911	5922	5933	5944	5955	5966	5977	5988	5999	6010	1 2 3	4 5 7	8 9 10
40	6021	6031	6042	6053	6064	6075	6085	6096	6107	6117	1 2 3	4 5 6	8 9 10
41	6128	6138	6149	6160	6170	6180	6191	6201	6212	6222	1 2 3	4 5 6	7 8 9
42	6232	6243	6253	6263	6274	6284	6294	6304	6314	6325	1 2 3	4 5 6	7 8 9
43	6335	6345	6355	6365	6375	6385	6395	6405	6415	6425	1 2 3	4 5 6	7 8 9
44	6435	6444	6454	6464	6474	6484	6493	6503	6513	6522	1 2 3	4 5 6	7 8 9
45	6532	6542	6551	6561	6571	6580	6590	6599	6609	6618	1 2 3	4 5 6	7 8 9
46	6628	6637	6646	6656	6665	6675	6684	6693	6702	6712	1 2 3	4 5 6	7 7 8
47	6721	6730	6739	6749	6758	6767	6776	6785	6794	6803	1 2 3	4 5 5	6 7 8
48	6812	6821	6830	6839	6848	6857	6866	6875	6884	6893	1 2 3	4 4 5	6 7 8
49	6902	6911	6920	6928	6937	6946	6955	6964	6972	6981	1 2 3	4 4 5	6 7 8

	0	1	2	3	4	5	6	7	8	9	1 2 3	4 5 6	7 8 9
50	6990	6998	7007	7016	7024	7033	7042	7050	7059	7067	1 2 3	3 4 5	6 7 8
51	7076	7084	7093	7101	7110	7118	7126	7135	7143	7152	1 2 3	3 4 5	6 7 8
52	7160	7168	7177	7185	7193	7202	7210	7218	7226	7235	1 2 2	3 4 5	6 7 7
53	7243	7251	7259	7267	7275	7284	7292	7300	7308	7316	1 2 2	3 4 5	6 6 7
54	7324	7332	7340	7348	7356	7364	7372	7380	7388	7396	1 2 2	3 4 5	6 6 7
55	7404	7412	7419	7427	7435	7443	7451	7459	7466	7474	1 2 2	3 4 5	5 6 7
56	7482	7490	7497	7505	7513	7520	7528	7536	7543	7551	1 2 2	3 4 5	5 6 7
57	7559	7566	7574	7582	7589	7597	7604	7612	7619	7627	1 2 2	3 4 5	5 6 7
58	7634	7642	7649	7657	7664	7672	7679	7686	7694	7701	1 1 2	3 4 4	5 6 7
59	7709	7716	7723	7731	7738	7745	7752	7760	7767	7774	1 1 2	3 4 4	5 6 7
60	7782	7789	7796	7803	7810	7818	7825	7832	7839	7846	1 1 2	3 4 4	5 6 6
61	7853	7860	7868	7875	7832	7889	7896	7903	7910	7917	1 1 2	3 4 4	5 6 6
62	7924	7931	7938	7945	7952	7959	7966	7973	7980	7987	1 1 2	3 3 4	5 6 6
63	7993	8000	8007	8014	8021	8028	8035	8041	8048	8055	1 1 2	3 3 4	5 5 6
64	8062	8069	8075	8082	8089	8096	8102	8109	8116	8122	1 1 2	3 3 4	5 5 6
65	8129	8136	8142	8149	8156	8162	8169	8176	8182	8189	1 1 2	3 3 4	5 5 6
66	8195	8202	8209	8215	8222	8228	8235	8241	8248	8254	1 1 2	3 3 4	5 5 6
67	8261	8267	8274	8280	8287	8293	8299	8306	8312	8319	1 1 2	3 3 4	5 5 6
68	8325	8331	8338	8344	8351	8357	8363	8370	8376	8382	1 1 2	3 3 4	4 5 6
69	8388	8395	8401	8407	8414	8420	8426	8432	8439	8445	1 1 2	3 3 4	4 5 6
70	8451	8457	8463	8470	8476	8482	8488	8494	8500	8506	1 1 2	3 3 4	4 5 6
71	8513	8519	8525	8531	8537	8543	8549	8555	8561	8567	1 1 2	3 3 4	4 5 5
72	8573	8579	8585	8591	8597	8603	8609	8615	8621	8627	1 1 2	3 3 4	4 5 5
73	8633	8639	8645	8651	8657	8663	8669	8675	8681	8686	1 1 2	3 3 4	4 5 5
74	8692	8698	8704	8710	8716	8722	8727	8733	8739	8745	1 1 2	3 3 4	4 5 5
75	8751	8756	8762	8768	8774	8779	8785	8791	8797	8802	1 1 2	3 3 4	5 5
76	8808	8814	8820	8825	8831	8837	8842	8848	8854	8859	1 1 2	3 3 4	5 5
77	8865	8871	8876	8882	8887	8893	8899	8904	8910	8915	1 1 2	3 3 4	4 5 5
78	8921	8927	8932	8938	8943	8949	8954	8960	8965	8971	1 1 2	3 3 4	4 5 5
79	8976	8982	8987	8993	8998	9004	9009	9015	9020	9025	1 1 2	3 3 4	4 5 5
80	9031	9036	9042	9047	9053	9058	9063	9069	9074	9079	1 1 2	3 3 4	4 5 5
81	9085	9090	9096	9101	9106	9112	9117	9122	9128	9133	1 1 2	3 3 4	4 5 5
82	9138	9143	9149	9154	9159	9165	9170	9175	9180	9186	1 1 2	3 3 4	4 5 5
83	9191	9196	9201	9206	9212	9217	9222	9227	9232	9238	1 1 2	3 3 4	4 5 5
84	9243	9248	9253	9258	9263	9269	9274	9279	9284	9289	1 1 2	3 3 4	4 5 5
85	9294	9299	9304	9309	9315	9320	9325	9330	9335	9340	1 1 2	3 3 4	4 5 5
86	9345	9350	9355	9360	9365	9370	9375	9380	9385	9390	1 1 2	3 3 4	4 5 5
87	9395	9400	9405	9410	9415	9420	9425	9430	9435	9440	0 1 1	2 2 3	3 4 4
88	9445	9450	9455	9460	9465	9469	9474	9479	9484	9489	0 1 1	2 2 3	3 4 4
89	9494	9499	9504	9509	9513	9518	9523	9528	9533	9538	0 1 1	2 2 3	3 4 4
90	9542	9547	9552	9557	9562	9566	9571	9576	9581	9586	0 1 1	2 2 3	3 4 4
91	9590	9595	9600	9605	9609	9614	9619	9624	9628	9633	0 1 1	2 2 3	3 4 4
92	9638	9643	9647	9652	9657	9661	9666	9671	9675	9680	0 1 1	2 2 3	3 4 4
93	9685	9689	9694	9699	9703	9708	9713	9717	9722	9727	0 1 1	2 2 3	3 4 4
94	9731	9736	9741	9745	9750	9754	9759	9763	9768	9773	0 1 1	2 2 3	3 4 4
95	9777	9782	9786	9791	9795	9800	9805	9809	9814	9818	0 1 1	2 2 3	3 4 4
96	9823	9827	9832	9836	9841	9845	9850	9854	9859	9863	0 1 1	2 2 3	3 4 4
97	9868	9872	9877	9881	9886	9890	9894	9899	9903	9908	0 1 1	2 2 3	3 4 4
98	9912	9917	9921	9926	9930	9934	9939	9943	9948	9952	0 1 1	2 2 3	3 4 4
99	9956	9961	9965	9969	9974	9978	9983	9987	9991	9996	0 1 1	2 2 3	3 3 4

CHEMICAL SCIENCE
PAPER III

Note :—(*i*) Question no. **1** is compulsory (**40** marks). Answer it in **800** words (8 pages).

(*ii*) Attempt any *ten* questions out of the remaining **35** questions (16 marks each). Answer each question in **300** words (3 pages).

1. (A) Answer any *three* of the following :

(*a*) By improving the instrument design the resolving power can be improved. However, a minimum width called 'natural line width' is inherent to all atomic and molecular transitions. What factors contribute to this ?

(*b*) TNT on detonation decomposes as

$$CH_3C_6H_2(NO_2)_{3\ (s)} \rightarrow \frac{7}{2}CO_{(g)} + \frac{3}{2}N_{2(g)} + \frac{5}{2}H_2O_{(l)} + \frac{7}{2}C_{(s)}$$

In a constant volume calorimeter, 454 mg of TNT at 25°C on detonation evolved 1.64 kcal. Neglecting volumes of solids and liquids, calculate ΔH and ΔU for the detonation. (MW of TNT = 227)

(*c*) Neatly sketch the possible fundamental modes of vibration for CO_2 molecule. Which of them are IR or Raman active vibrations ?

(*d*) The first rotational absorption of $^{12}C^{16}O$ is at 3.84235 cm^{-1} while that of $^{13}C^{16}O$ is at 3.67337 cm^{-1}. Taking the mass of oxygen to be 15.9994 and that of ^{12}C to be 12.00, calculate the atomic mass of ^{13}C. 12

(B) Write briefly on any *three* of the following :

(*a*) Superconducting Heterometal Oxides;

(*b*) Structural features of ZEOLITES make them fascinating materials in catalysis;

(*c*) Fischer-Tropsch process for Synthetic Gasoline;

(*d*) Unique behaviour Fluorine. 12

(C) Attempt any *three* of the following :

(*a*) A 100 MHz (23, 846 gauss) spectrometer observes a proton that absorbs at a field 0.20 gauss downfield from TMS. Determine its chemical shift and express this shift as a frequency.

(b) Propose a structure for compound that has the molecular formula $C_{10}H_{14}$ and shows the proton NMR signals at 7.1 δ(m, 5H) and 1.2(*s*, 9H).

(c) Deduce structural formula of compound X which shows : $M^+ = 117$, $\lambda_{max} \approx 234$ nm, ^{13}C-NMR shows six lines, IR—2230 cm^{-1}. The ^1H NMR signals are seen at 7.5 δ(*d*, 2H), 7.2 δ (*d*, 2H), 2.4 δ(*s*, 3H).

(d) In the mass spectrum of ethyl butanoate an intense peak at M/2 = 88 is seen. Explain its formation. 12

(D) Draw a block diagram of a double beam spectrophotometer. Name the source and the detectors used in uv, visible and IR regions. 4

2. (a) The lowest pH recorded experimentally so far is – 0.3 even in solutions with acid concentrations of greater than 10 N. Briefly give reasons.

(b) What is the function of Temperature control knob in a pH-meter ?

(c) For H_3PO_4, $K_1 = 7.5 \times 10^{-3}$, $K_2 = 6.2 \times 10^{-8}$ and $K_3 = 1.0 \times 10^{-12}$. If equal volumes of 0.1 N solutions of H_2NaPO_4 and HNa_2PO_4 solutions are mixed, what will be the pH of the mixture ?

3. (a) When common salt and crushed ice are mixed the temperature falls to – 21°C. Briefly explain this on the basis of phase rule.

(b) Draw Temperature composition diagram for Ethylalcohol-water system.

(c) For the reaction :

$$A \underset{k_{-1}}{\overset{k}{\rightleftharpoons}} B, \; k_1 = 10 \; s^{-1}$$

and the equilibrium constant is 20. Calculate the relaxation time for the reaction.

4. (a) Show that for a crystalline solid, the absolute entropy is given by

$$S_T = \int_0^T C_p d(ln \; T)$$

(*b*) A steam engine operates between 27°C and 100°C. Calculate its maximum possible efficiency. If the boiling point of water at a pressure of 50 atm is taken as 267°C (Superheated steam), what will be the efficiency of the engine ? In both cases the temperature of the sink is 27°C.

(*c*) Carbonated beverages are bottled with CO_2 under high pressure. Why does the mixture fizz when the container is suddenly opened to the atmosphere ? Use Le Chatelier's principle to describe what happens ?

5. (*a*) Given $du = T.ds - P.dv$, do a Legendre transformation to get the thermodynamic quantity A.

(*b*) Critical micelle concentration of an aqueous solution of sodium dodecyl sulfate is 10^{-3} mol. lit^{-1} at 25°C. What is $\Delta G°$ for micellization ?

(*c*) (*i*) Which will have higher aggregation number : Lithium dodecyl sulfate a sodium dodecyl sulfate ?

(*ii*) Which will show higher CMC : CTAB or PEO having equal number of carbon atoms ?

6. (*a*) 'Stern model for electrical double layer is superior to Gouy-Chapman model'. Give reasons.

(*b*) Draw Evan's diagram to show inhibition of corrosion.

(*c*) The standard reduction potential of Cu and Ag are + 0.34 V and + 0.79 V respectively. Calculate the e.m.f. of the cell formed by their combination. Indicate how the cell can be represented. Write the electrode and cell reactions.

7. (*a*) A finely divided metal shows high catalytic activity but it is difficult to handle it. Suggest an alternative to maintain high activity and name characterization techniques you may use.

(*b*) Write BET equation, draw BET plot and show how monolayer volume can be determined from it.

(*c*) Why physisorption has no catalytic application ?

8. (*a*) The resonance reabsorption of γ-rays is difficult to observe. What conditions Mössbauer used to realize it ?

(*b*) State the conditions for molecules to be microwave active and infrared active.

(*c*) What are magnetogyric ratio and the g_N factor of the ^{31}P nucleus, the magnetic moment of which is 1.1305 μ_N. (Given $\mu_N = 5.051 \times 10^{-27}$ J . T^{-1} and $h = 6.626 \times 10^{-34}$ Js).

9. (*a*) In what respect high temperature fuel cell is better than low temperature fuel cell ?

(*b*) Give important differences between step growth polymerization and chain growth polymerization.

(*c*) In a given space there are 20 particles. If an imaginary wall divides space into two parts, compute the probability that all the 20 particles are on the same side of the wall.

10. (*a*) The entropy is defined as

$$S = -k\sum_i P_i \ln P_i$$

where $P_i = e^{-\beta E_i}/Q$ and $Q = \sum_i e^{-\beta E_i}$. Explain S in terms of Q.

(*b*) Show that statistical mechanical partition function has same property as quantum mechanical wave function.

(*c*) Air consists of a mixture of 21% of oxygen and 79% N_2 by volume. Calculate the entropy of mixing of the elements of air.

11. (*a*) Show that an oscillating dipole has frequency components $\nu \pm \nu_{vib}$ as well as the exciting frequency ν.

(*b*) From the following data comment on the molecular structure of nitrous oxide and assign the observed lines to specific molecular vibrations :

$\bar{\nu}$ (cm^{-1})	IR	Raman
589	strong, PQR Contour	—
1285	very strong, PR	very strong, polarized
2224	very strong, PR	strong, depolarized

(*c*) What do you mean by 'hot bands' in spectroscopy ?

12. (*a*) Give the valence electron count for the following species. Which one/s conforms to the E.A.N. rules ?

(*i*) $Co_2(\mu_2\text{-}CO)_2(CO)_6$

(*ii*) $W(CO)_6$

(*iii*) $Cr(CNMe)_6$

(*iv*) $[Fe(CN)_6]^{4-}$

(*v*) $[Cr(CO)_5]^{2-}$

(*vi*) $[Mn(CO)_5(CH_2C_6H_5)]^-$

(*vii*) $Fe(CO)_4Br_2$

(*viii*) $[Mn(CO)_5]^{2-}$

(*ix*) $Ni(PPh_3)_4$

(*x*) $H\ Rh(CO)_4$

(*xi*) $Ru_3(CO)_{12}$

(*xii*) $[Co(CN)_5]^{2-}$.

(*b*) Name each one of the above species.

(*c*) Using Wilkinson's Complex, discuss the Homogeneous Catalytic Hydrogenation of Propene.

13. Write briefly on :

(*a*) Structure and Role of Biological Membranes during ion transport.

(*b*) Biological Nitrogen Fixation.

(*c*) Icosahedral structure of $[B_{12}H_{12}]^{2-}$ anion.

(*d*) Structural features and Correlation between the structures of Heme group proteins.

14. (*a*) (*i*) Sketch π bonding orbitals that result from the combination of the following orbitals on separate atoms : p_x and p_x ; p_z and d_{xz} ; and d_{xz} and d_{xz}.

(*ii*) Write the molecular orbital configuration diagrams and give the bond order of NO^+ ; NO_2 and NO^-. Which one/s of these species should be paramagnetic ?

(*b*) Give molecular orbital diagram of COBALTOCENE and explain its magnetic behaviour.

15. (*a*) How one can distinguish between the following isomers ?

(*i*) $[Co(NH_3)_5Br]SO_4$ and $[Co(NH_3)_5SO_4]Br$

(*ii*) $[Co(NH_3)_3(NO_2)_3$ and $[Co(NH_3)_6]$ $[Co(NO_2)_6]$

(*iii*) Cis-and Trans $[CoCl_2(en)_2]Cl$

(*iv*) Cis-and Trans $[Pt(glycine)_2]$

(*v*) Cis-and Trans $NH_4[Co(NO_2)_4$ $(NH_3)_2]$

(*b*) Cite one example each of the following types of isomerism :

(*i*) Hydration Isomerism;

(*ii*) Coordination Isomerism;

(*iii*) Linkage Isomerism;

(*iv*) Ionization Isomerism;

(*v*) Geometrical Isomerism;

(*vi*) Optical Isomerism.

(*c*) Measurement of 10 Dq and discuss the factors responsible for the magnitude of 10 Dq.

16. Write briefly on :

(*a*) Ziegler-Natta Catalysed Polymerization of ethylene.

(*b*) Water gas shift reaction using Homogeneous catalyst.

(*c*) Fluxional behaviour of Pentacoordinated molecule.

(*d*) ZSM-5, its nature and applications.

17. (*a*) Arrange the following in the increasing order of property given in each case :

(*i*) Oxidizing capacity of CrO_4^{2-}, FeO_4^{2-} and $[VO_4]^{3-}$

(*ii*) First ionization energy of S, Al, P and Mg

(*iii*) N–N bond order in N_2, N_2F_2 and N_2H_4

(*iv*) Ionic radii O^{2-}, N_3^- and O_2^-.

(*b*) (*i*) In solid state CuF_2, the Cu—F distance is found at 1.93 Å (four bonds) and two bonds of Cu—F at 2.27 Å.

(*ii*) Magnetic properties of $[Fe(H_2O)_6]^{3+}$ and $[Fe(CN)_6]^{3-}$, give their comparison.

(*iii*) Explain why solid ferrocene exhibit staggered conformation. While ferrocene in vapour form possesses Eclipsed conformation.

(*iv*) Back donation in metal alkene and metal alkyne complexes stabilize these compounds. Discuss in brief.

18. (*a*) Write the highest coordination number exhibited by the following ligands :

(*i*) Acetyl acetone;

(*ii*) Porphyrin;

(*iii*) 18 Crown-6 ether;

(*iv*) Diethylene triamine;

(*v*) Diethylene triamine penta-acetic acid;

(*vi*) Valinomycin;

(*vii*) Schiff base Salen;

(*viii*) EDTA;

(*ix*) Triphenyl phosphine;

(*x*) Tripeptide glutatheone (γL glutamyl Lcysteinyl glycine)

(*b*) Discuss the electronic configuration of lanthanides and lanthanide contraction.

19. (*a*) (*i*) Give the structural formulae of two metallocenes with bent structures

(*ii*) Structure of σ bonded metal alkyne and metal allylene compounds.

(*iii*) Factors governing the Ionization energy should be discussed in brief.

(*iv*) Give four points of differences between Boron and Aluminium.

(*v*) Why CO_2 is a gas while SiO_2 is solid ?

(*b*) What are Silicones ? How are these prepared and discuss in brief their application ?

20. (*a*) Give Schematic representation of Homogeneously Catalysed Hydroformylation of Ethylene using Cobalt Carbonyl.

(*b*) A black substance (X) when heated in presence of KNO_3 and KOH produced a green mass (Y). On extraction with water Y gave a green solution. The green solution turned purple (Z) when a current of chlorine was passed through it.

(*c*) What are porphyrins and discuss in brief the Biogenesis of Porphyrin ?

21. (*a*) What are Ionophores and what role these play in understanding the biochemistry of alkali metals ?

(*b*) Discuss critically but in brief the reversibly binding of dioxygen to Haemoglobin.

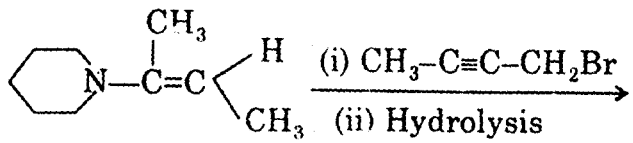
(*c*) General structure of chlorophyll and discuss how chlorophyll facilitate photosynthesis.

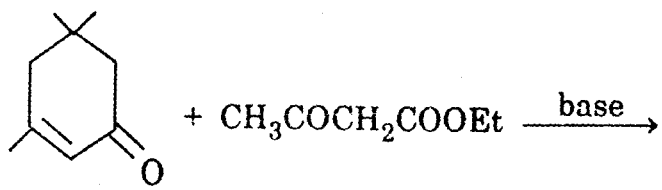
22. (*a*) A classical method of synthesizing α-amino acids is based upon modified phthalimido malonic ester method. Utilizing this concept suggest a synthesis of glutamic acid. 8

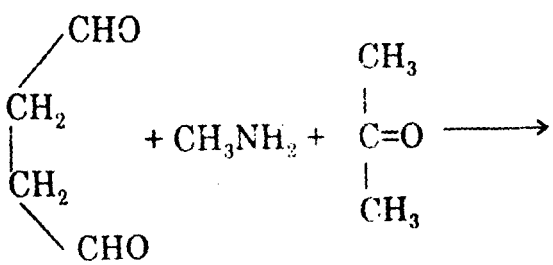
(*b*) Allose is an aldohexose with a structure similar to glucose except for the opposite stereochemical configuration at C_3. Draw Haworth structure for a β-anomer of D-allose. 4

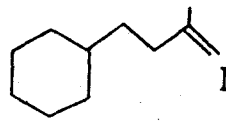
(*c*) Treatment of 3-chlorocyclopropene with $SbCl_5$ yields a stable crystalline solid (I) of formula $C_3H_3SbCl_6$. I is insoluble in non-polar solvents but soluble in polar solvents. The ^1H NMR shows three exactly equivalent protons. Give most likely structure of I and show how this structure accounts for the various observations. 4

23. Predict the product and name the reactions involved wherever possible (any *four*) :

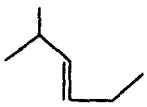
(*i*) 

(*ii*) 

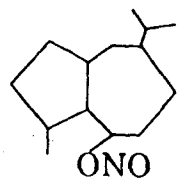
(*iii*) 

(*iv*) 

(i) Bu^n Li in the presence of
(i) $(CH_3)_2N.CH_2.CH_2.N(CH_3)_2$
(ii) D_2O

(*v*) 

L(+) – diethyl tartrate
$Ti(OPri)_4$ + Bu^tOOH + CH_2Cl_2; –20°C

(*vi*) 

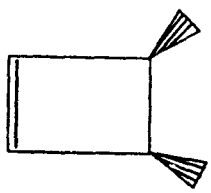
$h\nu$

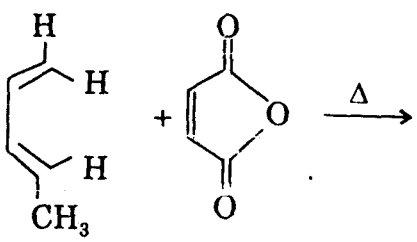
16

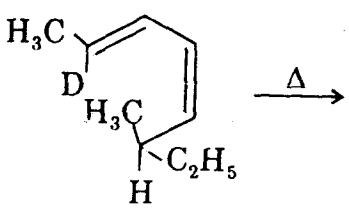
24. (A) In the following pericyclic reactions :

(*i*) Predict the product and its stereochemistry; and

(*ii*) Name the reaction type and give selection rules (attempt any *two*) :

(*a*) 
175°C

(*b*) 

(*c*) 

10

(B) There are three isomeric pyridine mono-carboxylic acids :

D, m.p. 137°C, E, m.p. 234–237°C, F, m.p. 317°C

Their structures were proved as follows :

Quinoline + $KMnO_4$ $\xrightarrow{OH^-}$ a dibasic acid ($C_7H_5NO_4$) $\longrightarrow$

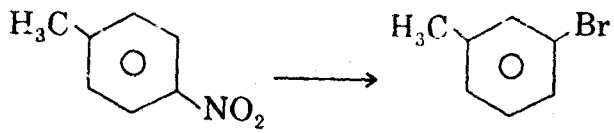
$\xrightarrow{\Delta}$ E(m.p. 234 – 237°C)

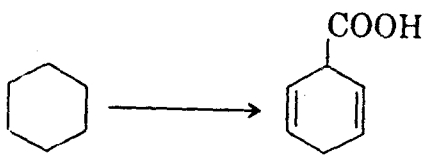
Isoquinoline + $KMnO_4$ $\xrightarrow{OH^-}$ a dibasic acid ($C_7H_5NO_4$) $\longrightarrow$

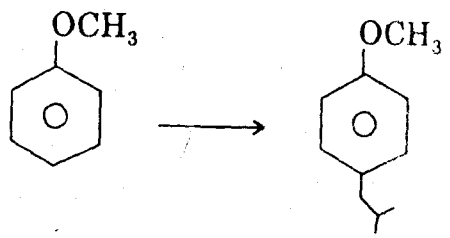
$\xrightarrow{\Delta}$ E(m.p. 234 – 237°C) + F(m.p 317°C)

What structure should be assigned to D, E and F ? 6

25. Describe the reagents and the intermediates required for each of the following conversions (more than one step may be required). Attempt any *two* :

(*i*) 

(*ii*) 

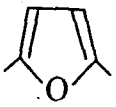
(*iii*) 

16

26. Predict the product of the following reactions (any *two*) :

(*i*)  $+ (CH_3CO)_2O \xrightarrow[0°C]{BF_3}$ product A $\xrightarrow[\text{(ii) } H_3O^+]{\text{(i) } I_2 + NaOH_{(aq)},\ 0°C}$

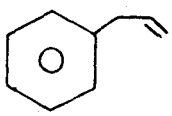
product B $\xrightarrow[100°C]{Br_2}$ product C

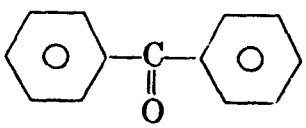
(*ii*)  $\xrightarrow{H_2SO_4(aq)}$ product A $\xrightarrow{C_6H_5CH_2NH_2}$ product B

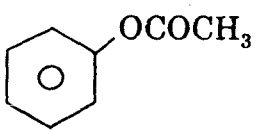
(*iii*)  $\xrightarrow{\text{per-acid}}$ product A $\xrightarrow[H_2SO_4]{HNO_3}$ product B $\xrightarrow{PCl_3}$ product C

16

27. (*a*) Predict the product and suggest a mechanism of the following photochemical transformations (any *two*) :

(*i*)  $\xrightarrow{h\nu}$

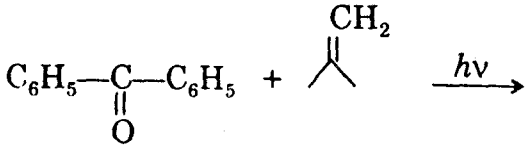
(*ii*)  $\xrightarrow[(CH_3)_2CH.OH]{h\nu}$

(*iii*)  $\xrightarrow{h\nu}$

12

(*b*) Complete the following reactions :

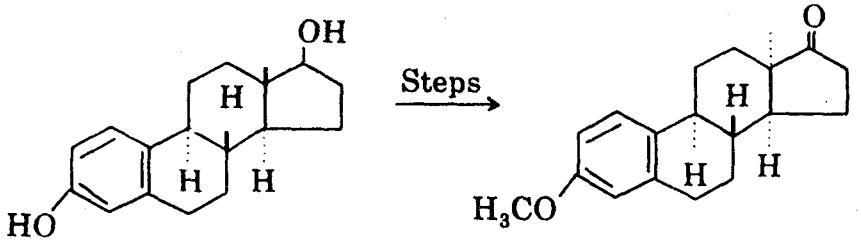
(*i*)  $\xrightarrow[\text{Photosensitizer}]{h\nu}$

(*ii*) 

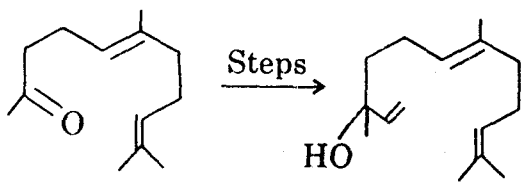
4

28. (*a*) How will you bring about the following conversions using appropriate reagents/reactions ?

(*i*)



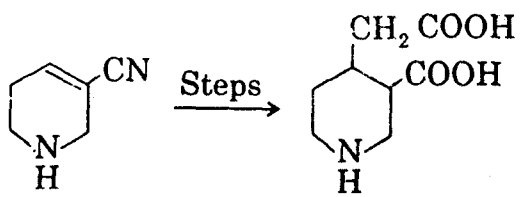
(*ii*)



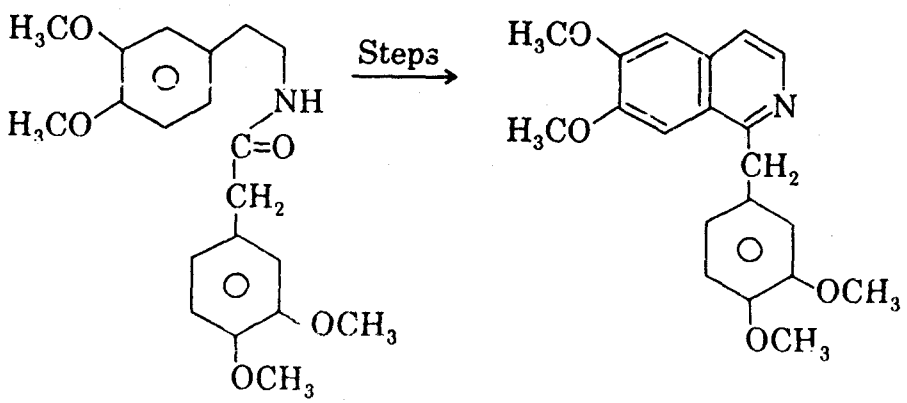
10

(*b*) How will you perform the following conversions ?

(*i*)

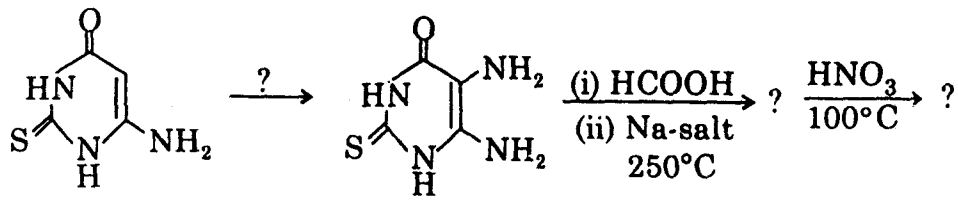


(*ii*)



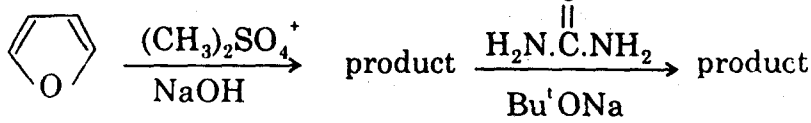
6

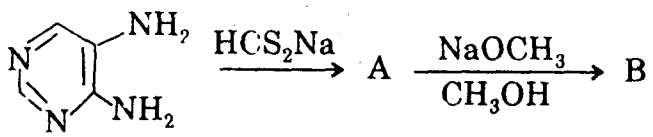
29. (*a*) Complete the following reations :



9

(*b*) Attempt any *one* of the two reactions :

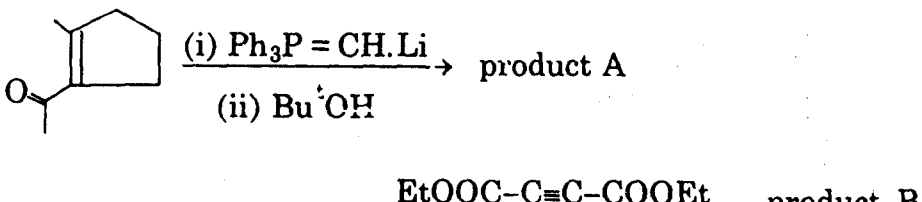
(*i*) 

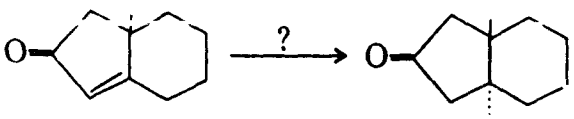
(*ii*) 

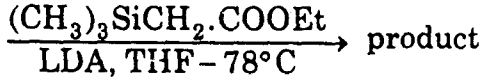
4

(*c*) Suggest the biogenesis of β-pinene starting from geranyl pyrophosphate. 3

30. (*a*) Complete the following reaction sequences :

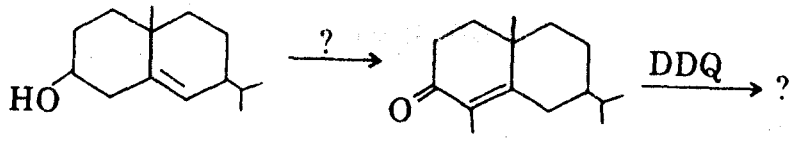
(*i*) 

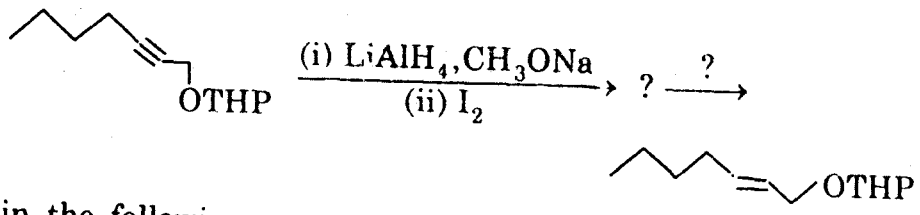
(*ii*) 



10

(*b*) Complete the following reactions :

(*i*) 

(*ii*) 

6

31. (*a*) Explain the following :

(*i*) Cyclohexanone $\xrightarrow[\text{acid}]{\text{metal}+}$ equatorial alcohol

Cyclohexanone $\xrightarrow{[(CH_3)_2CHO]_3Al}$ axial alcohol.

(*ii*) On heating cis-4-hydroxycyclohexane Carboxylic acid forms a lactone; but trans-4-hydroxy cyclohexane carboxylic acid does not. Draw the stereo-structures of two carboxylic acids and lactone.

8

(*b*) Given below in column A are some common organic compounds while column B lists a possible use for each of the compound. Match the entries in the two columns by correctly identifying the use for each of the compounds :

A		B	
(*a*)	Ascorbic acid	(*i*)	A nucleotide
(*b*)	Adenylic acid	(*ii*)	A drug intermediate
(*c*)	Anthranilic acid	(*iii*)	A vitamin
(*d*)	Salicylic acid	(*iv*)	A dye intermediate

4

(*c*) How is genetic code transcripted in protein biosynthesis ? 4

32. (*a*) Why are hollow cathode lamps less prone to self absorption than electrodeless discharge lamps ?

(*b*) 0.0121 g of benzoic acid (Mol wt 121) was dissolved in 100 ml of methanol water (1 : 1). If a 10 ml aliquot of this solution was further diluted to 100 ml which showed an absorbance of 0.80 at 265 nm in a 10 mm cell, what is the molar absorptivity ?

(*c*) Draw a schematic diagram of DTA. How is it different from DSC ?

33. (*a*) What could be the shape of the titration curve for the photometric titration of bismuth (III) with EDTA at 265 nm.

(*b*) Why should monochromators contain as few reflecting surfaces as possible ? What is the relationship between a monochromator's bandwidth and the width of a spectral feature which ensures that it will be accurately recorded ?

(*c*) What are the causes of background inteferences in AAS ?

34. (*a*) Why are generally group V (As, Sb) elements analysed by the hydride generation method in AAS ?

(*b*) When a simple digestion by heating a sample with acid or base in the hood does not dissolve the sample that is required for analysis, a digestion with a flux or microwave digestion in a closed container may solve the problem. What are the advantages of each of these techniques over simple acid digestion ?

(*c*) How is thermogravimetry used for the quantitative analysis of a mixture of calcium and strontium ?

35. (*a*) What are the factors on which glass transition temperature of a polymer depend ? How is T_g determined ?

(*b*) A sample of L-dopa is electrolysed exhaustively with 42.0 μC. The process requires two electrons per molecule. The formula weight of L-dopa is 197.2. What is the mass of L-dopa in the sample ?

(*c*) How is UV-visible spectrometry used for the analysis of mixtures of absorbing substances ?

36. (*a*) Why is linear scan voltammograms generally assume a sigmoidal curve ?

(*b*) Why are stripping methods more sensitive than other voltammetric procedures ?

(*c*) Why is square wave polarography of great speed and high sensitivity ?

Q. No. .....